

What is the energy gained in the H ₂ O when you melt 326 g of ice into liquid?	What is the energy gained in the H ₂ O when you melt 326 g of ice into liquid?	What is the energy gained in the H ₂ O when you melt 326 g of ice into liquid?	What is the energy gained in the H ₂ O when you melt 326 g of ice into liquid?	What is the energy gained in the H ₂ O when you melt 326 g of ice into liquid?
What is the energy absorbed when you heat 326 grams of water from 0°C to 100°C?	What is the energy absorbed when you heat 326 grams of water from 0°C to 100°C?	What is the energy absorbed when you heat 326 grams of water from 0°C to 100°C?	What is the energy absorbed when you heat 326 grams of water from 0°C to 100°C?	What is the energy absorbed when you heat 326 grams of water from 0°C to 100°C?
What is the energy needed to melt 326 grams of ice and heat it to 100 °C?	What is the energy needed to melt 326 grams of ice and heat it to 100 °C?	What is the energy needed to melt 326 grams of ice and heat it to 100 °C?	What is the energy needed to melt 326 grams of ice and heat it to 100 °C?	What is the energy needed to melt 326 grams of ice and heat it to 100 °C?
Determine the energy required to convert 21.1 grams of ice at -6°C to steam at 100°C	Determine the energy required to convert 21.1 grams of ice at -6°C to steam at 100°C	Determine the energy required to convert 21.1 grams of ice at -6°C to steam at 100°C	Determine the energy required to convert 21.1 grams of ice at -6°C to steam at 100°C	Determine the energy required to convert 21.1 grams of ice at -6°C to steam at 100°C
What is the heat required to convert 51 grams of ice at -20.3°C to water at 0°C?	What is the heat required to convert 51 grams of ice at -20.3°C to water at 0°C?	What is the heat required to convert 51 grams of ice at -20.3°C to water at 0°C?	What is the heat required to convert 51 grams of ice at -20.3°C to water at 0°C?	What is the heat required to convert 51 grams of ice at -20.3°C to water at 0°C?
What is the energy absorbed when you melt 75 grams of ice at -5°C to steam at 150°C?	What is the energy absorbed when you melt 75 grams of ice at -5°C to steam at 150°C?	What is the energy absorbed when you melt 75 grams of ice at -5°C to steam at 150°C?	What is the energy absorbed when you melt 75 grams of ice at -5°C to steam at 150°C?	What is the energy absorbed when you melt 75 grams of ice at -5°C to steam at 150°C?